## organic papers

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#### **Key indicators**

Single-crystal X-ray study T = 293 K Mean  $\sigma$ (C–C) = 0.003 Å Disorder in main residue R factor = 0.057 wR factor = 0.160 Data-to-parameter ratio = 14.6

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

# 4-Cyano-3-fluorophenyl (2*E*)-3-(4-heptyloxy-3-methoxyphenyl)acrylate

In the title compound,  $C_{24}H_{26}FNO_4$ , the two benzene rings are almost perpendicular to one another, with a dihedral angle of 87.7 (1)°. Molecules are linked into two-dimensional layers by non-classical intermolecular hydrogen-bonding interactions. The packing is further stabilized by  $\pi$ - $\pi$  interactions.

## Comment

We have reported the structure of 4-cyano-3-fluorophenyl 4'isobutoxybiphenyl-4-carboxylate, (II) (Ren *et al.*, 2005). In our ongoing research into liquid-crystal materials containing fluorine, the title compound, (I), was obtained and its structure is reported here.



Bond lengths and angles in (I) are within normal ranges (Allen *et al.*, 1987), and the values in the 4-cyano-3-fluorophenyl group are comparable with those in the related structure 4-cyano-3-fluorophenyl 4'-isobutoxybiphenyl-4-carboxylate (Ren *et al.*, 2005). The 4-heptyloxy-3-methoxyphenyl group, excluding atom C24, is essentially coplanar with the acrylate fragment. This combined unit is almost perpendicular to the 4-cyano-3-fluorophenyl group, with a dihedral angle of 87.7 (1)° between the two benzene rings. The C8–C9–C10–C11 torsion angle is 178.9 (2)°, indicating a *trans* configuration of the double bond. The F atom shows disorder over two positions, with site occupancies of 0.638 (4) and 0.362 (4).

There is one intramolecular hydrogen-bond interaction,  $C10-H10\cdots O1$ , forming a five-membered ring, which contributes to the planarity of the acrylate group. In the crystal structure, molecules are linked into two-dimensional



### Figure 1

The structure of the compound (I), showing 50% probability displacement ellipsoids and the atom-numbering scheme. The fluorine is disordered over two positions. Both components are shown.

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+ 0.2038P] where  $P = (F_0^2 + 2F_c^2)/3$ 

 $\Delta \rho_{\rm min} = -0.21 \text{ e } \text{\AA}^{-3}$ 

 $(\Delta/\sigma)_{\rm max} < 0.001$  $\Delta \rho_{\rm max} = 0.23 \text{ e} \text{ Å}^{-3}$ 



Figure 2

A view, down the b axis, of the packing. Hydrogen bonds are indicated by dashed lines.

layers by intermolecular C9-H9···O2<sup>i</sup> and C17-H17B···F<sup>ii</sup> interactions [symmetry codes: (i) 2 - x, 1 - y, 1 - z; (ii) x - 1, y, 1 + z] (Fig. 2 and Table 2). The packing is further stabilized by  $\pi - \pi$  interactions between the benzene (C2–C7) rings, the distance between the centroids,  $Cg \cdots Cg^{iii}$  [symmetry code: (iii) 2 - x, -1 - y, -z], being 3.713 Å.

## **Experimental**

To a solution of 3-(4-hydroxy-3-methoxyphenyl)acrylic acid (1.94 g, 0.01 mol) in 17.0% KOH solution (10 ml) was added dropwise 1bromoheptane (0.54 g, 0.03 mol) in ethanol (30 ml). The mixture was heated under reflux for 24 h and extra KOH solution was added to remove the by-products. The mixture was then acidified to give a white precipitate, which was filtered and recrystallized from ethanol, giving a white precipitate, identified as 4-hydroxy-3-methoxyphenylacrylic acid. This compound (1.46 g) and DCC (1,3-dicyclohexylcarbodiimide, 0.6 g) were dissolved in THF (20 ml), and a solution of 2-fluoro-4-hydroxybenzonitrile (0.68 g) and DMAP (4dimethylaminopyridine, 0.01 g) in THF (20 ml) was added. The mixture was stirred at room temperature for 24 h and then filtered. Colorless single crystals of (I) suitable for X-ray diffraction study were obtained by slow evaporation of an ethyl acetate/petroleum ether (1:4 v/v) solution over a period of 2 d.

#### Crystal data

$C_{24}H_{26}FNO_4$	Z = 2
$M_r = 411.46$	$D_x = 1.207 \text{ Mg m}^{-3}$
Triclinic, $P\overline{1}$	Mo $K\alpha$ radiation
$a = 7.763 (4) \text{ Å}_{-}$	Cell parameters from 2069
b = 11.207 (5)  Å	reflections
c = 13.631 (7)  Å	$\theta = 2.5 - 25.9^{\circ}$
$\alpha = 82.499 \ (8)^{\circ}$	$\mu = 0.09 \text{ mm}^{-1}$
$\beta = 82.269 \ (8)^{\circ}$	T = 293 (2) K
$\gamma = 75.662 \ (8)^{\circ}$	Column, colorless
$V = 1132.6 (10) \text{ Å}^3$	$0.31 \times 0.30 \times 0.14 \text{ mm}$

#### Data collection

Siemens SMART 1000 CCD area-	4124 independent reflections
detector diffractometer	3035 reflections with $I > 2\sigma(I)$
$\omega$ scans	$R_{\rm int} = 0.014$
Absorption correction: multi-scan	$\theta_{\rm max} = 25.5^{\circ}$
(SADABS; Sheldrick, 1996)	$h = -9 \rightarrow 8$
$T_{\min} = 0.973, \ T_{\max} = 0.988$	$k = -13 \rightarrow 13$
5944 measured reflections	$l = -14 \rightarrow 16$
Refinement	
Refinement on $F^2$	$w = 1/[\sigma^2(F_o^2) + (0.0764P)^2]$

 $\begin{array}{l} R[F^2>2\sigma(F^2)]=0.057\\ wR(F^2)=0.160 \end{array}$ S = 1.064124 reflections 283 parameters H-atom parameters constrained

## Table 1

Selected bond lengths (Å).

F-C7	1.329 (3)	O3-C17	1.423 (2)
F'-C3	1.291 (5)	O4-C14	1.366 (2)
O1-C8	1.369 (2)	O4-C18	1.439 (2)
O1-C5	1.393 (2)	N-C1	1.144 (3)
O2-C8	1.193 (2)	C8-C9	1.457 (3)
O3-C13	1.365 (2)	C9-C10	1.332 (3)

Table 2 Hydrogen-bond geometry (Å, °).

$D - H \cdot \cdot \cdot A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdots A$
$C9-H9\cdots O2^{i}$ $C10-H10\cdots O1$ $C17-H178\cdots F^{ii}$	0.93	2.58	3.509 (3)	177
	0.93	2.35	2.730 (3)	104
	0.96	2.51	3.211 (4)	130

Symmetry codes: (i) -x + 2, -y + 1, -z + 1; (ii) x - 1, y, z + 1.

All H atoms except H3 and H7 were located in difference Fourier maps and constrained to ride on their parent atoms, with C-H distances in the range 0.93–0.97 Å, and with  $U_{iso}(H) = 1.2U_{eq}(C)$  or  $U_{\rm iso}({\rm H}) = 1.5 U_{\rm eq}({\rm methyl~C}).$ 

Data collection: SMART (Siemens, 1996); cell refinement: SAINT (Siemens, 1996); data reduction: SAINT; program(s) used to solve structure: SHELXTL (Sheldrick, 1997); program(s) used to refine structure: SHELXTL; molecular graphics: SHELXTL; software used to prepare material for publication: SHELXTL, PARST (Nardelli, 1995) and PLATON (Spek, 2003).

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